Improved Rotational Constants for HF1

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The ground state rotational constants for HF were recently reported (1). These molecular constants were determined from spectral measurements that covered the spectral range from 40 to 1100 cm⁻¹ (1 to 32 THz). The high-J transitions were measured by a comparison with known spectral lines and had uncertainties in the predicted line frequencies of as much as 100 MHz (2σ).

We have remeasured the $J=27 \leftarrow 26$ and $J=33 \leftarrow 32$ transitions using a tunable diode laser heterodyne spectroscopy technique (2) with frequencies of 28 666 999 \pm 8 and 32 557 132 \pm 8 MHz, respectively. The uncertainty of 8 MHz (2 σ) is 9 times less than the 72 MHz reported in Ref. (1). The HF was prepared in a 1-cm-diameter 1.2-m-long water-cooled cell by a 30 mA dc discharge in a 66 Pa 10:1 mixture of He:F and 13 Pa of H₂. The frequency of the diode laser was measured by heterodyning with a stabilized CO₂ laser.

Using the new frequencies along with the previously determined transition frequencies, we have redetermined the rotational constants as defined by the equation for the energy levels (3):

$$E_{v}(J) = B_{v}J(J+1) - D_{v}[J(J+1)]^{2} + H_{v}[J(J+1)]^{3} - L_{v}[J(J+1)]^{4} + M_{v}[J(J+1)]^{5}.$$

These new values for the ground state rotational constants of HF are listed in Table I. The improvement in

TABLE I

Rotational Constants of Ground State HF

	MHz*	МНz ^b
Ba	616365.200(10)	616365.199(75)
D ₀	63.5524(4)	63.5532(41)
но	4.898(2)10-3	4.897(15)10-3
Lo	4,44(2)10-7	4.41(20)10-7
Ma	2.94(8)10-11	2.82(80)10-11

The uncertainty in the last digits (2σ) is given in parentheses.

- a. This work.
- b. Ref. 1.

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accuracy of these two transition frequencies shows up as a reduction in the uncertainty in the molecular constants by nearly an order of magnitude in every case.

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REFERENCES

- D. A. Jennings, K. M. Evenson, L. R. Zink, C. Demuynck, J. L. Destombes, B. Lemoine, and J. W. C. Johns, J. Mol. Spectrosc. 122, 477–480 (1987).
- 2. J. S. Wells, F. R. Petersen, A. G. Maki, and D. J. Sukle, Appl. Opt. 20, 1676-1684 (1981).
- 3. W. GORDY AND R. L. COOK, "Microwave Molecular Spectra." Wiley, New York, 1984.